

Falsifying some conjectures in organic chemistry on the structure of perfect matchings in subgraphs of the honeycomb lattice

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Abstract

In mathematical chemistry, we abstract the *carbon skeleton* of a molecular structure as an undirected graph, where vertices represent carbon atoms, edges represent single carbon-carbon bonds, and we ignore the presence of other atom types (e.g., hydrogen). Here, a *polyhex* is a connected carbon skeleton that is isomorphic to an induced subgraph of the honeycomb lattice with minimum degree 2. If a polyhex has at most one face (respectively, more than one face) of length ≥ 7 , corresponding to the outer boundary (respectively, outer boundary or a hole), then we may refer to it as a *benzenoid* (respectively, a *coronoid*). Letting G be a polyhex, we refer to a perfect matching for G as a *Kekulé structure*, and for a particular Kekulé structure \mathcal{M} , we refer to any 6-sided face in G with 3 edges in \mathcal{M} as a *resonant benzene face*. We can now define the *Fries number* $F(G)$ [4] for G as the maximum number of resonant benzene faces that can be realized by any Kekulé structure. Additionally, we can define the *Clar number* $C(G)$ [2] for G as the maximum number of mutually vertex-disjoint resonant benzene faces that can be realized by any Kekulé structure. Finally, we can define the *Fries set* \mathcal{F} (respectively, *Clar set* \mathcal{C}) for G as the set of all Kekulé structures inducing $F(G)$ resonant benzene faces (respectively, $C(G)$ mutually vertex-disjoint resonant benzene faces).

In this talk, we will elaborate on coronoid counterexamples we have discovered for the well-known *Clar–Fries conjecture* [1,3,5,6,7] for polyhexes (or bipartite planar carbon skeletons more generally [1]) that at least one Clar set is a subset of a Fries set. It was previously only known that the Clar–Fries conjecture was false for fullerenes [6,7] (i.e., necessarily non-bipartite 3-regular planar graphs with pentagonal and hexagonal faces), and true for special classes of fullerenes [6,7] and benzenoids [5].

References

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